Tetrahedron 65 (2009) 7422–7428

Contents lists available at [ScienceDirect](www.sciencedirect.com/science/journal/00404020)

Tetrahedron

journal homepage: www.elsevier.com/locate/tet

Versatile synthesis of epicatechin series procyanidin oligomers, and their antioxidant and DNA polymerase inhibitory activity $\dot{\mathbf{r}}$

Akiko Saito $^{\mathrm{a}_{\ast},\dagger}$, Yoshiyuki Mizushina $^{\mathrm{b},\mathrm{c}}$, Akira Tanaka $^{\mathrm{d}}$, Noriyuki Nakajima $^{\mathrm{a},\mathrm{e},\ast}$

^a Biotechnology Center, Toyama Prefecture, Japan

^b Laboratory of Food & Nutritional Sciences, Department of Nutritional Science, Kobe-Gakuin University, Nishi-ku, Kobe, Hyogo 651-2180, Japan

^c Cooperative Research Center of Life Sciences, Kobe-Gakuin University, Nishi-ku, Kobe, Hyogo 651-2180, Japan

^d Department of Biosources Science, College of Technology, Toyama Prefectural University, Japan

e Department of Biotechnology, Faculty of Engineering, Toyama Prefectural University, Imizu, Toyama 939-0398, Japan

article info

Article history: Received 2 June 2009 Received in revised form 4 July 2009 Accepted 6 July 2009 Available online 9 July 2009

Keywords: Proanthocyanidin Condensed tannin Epicatechin Catechin Antioxidant Radical scavenging activity DNA polymerase inhibitory activity

ABSTRACT

Proanthocyanidins, known as condensed tannins or oligomeric flavonoids, exist in many edible plants and show various interesting biological activities. We have developed a simple and versatile method of synthesizing procyanidin oligomers consisting of $(-)$ -epicatechin and $(+)$ -catechin. This method is applicable to the synthesis of various 3-O-substituted oligomers. We report here the stereoselective and length controlled synthesis of [4–8]-condensed (-)-epicatechin series procyanidin oligomers. We described the details of the synthesis of an two tetramers, $(-)$ -epicatechin- $(-)$ -epicatechin- $(-)$ -epicatechin-(-)-epicatechin and (-)-epicatechin-(-)-epicatechin-(-)-epicatechin-(+)-catechin (arecatannin A1), (-)-epicatechin pentamer and two 3,3",3""-tri-O-galloyl trimers, (-)-epicatechin-(-)-epicatechin-(-)-epicatechin-3,3",3""-tri-O-gallate and (-)-epicatechin-(-)-epicatechin-(+)-catechin-3,3",3""-tri-Ogallate with the condensation method using TMSOTf as a catalyst. The ability of DPPH radical scavenging activity and DNA polymerase inhibitory activity of these oligomeric compounds were investigated.

- 2009 Elsevier Ltd. All rights reserved.

Tetrahedror

1. Introduction

Oxygen is biologically important for energy production, but active oxygen and free radicals in the body can injure cells and genes. They are also thought to be a cause of cancer, a lifestyle-related disease, and to contribute to aging. Therefore, there is currently great interest in the research and investigation into compounds that have strong anti-oxidation activity and superior ability to scavenge radicals. Thus, food and ingredients that can eliminate active oxygen and free radicals have recently received increased attention.

Proanthocyanidins (condensed tannins or oligomeric flavo-noids)^{[1,2](#page-6-0)} are known to be extremely strong antioxidants: investigating them has become increasingly important because of various strong biological activities. In many cases, however,

E-mail addresses: saiaki@riken.jp (A. Saito), nori@pu-toyama.ac.jp (N. Nakajima). Present address: Antibiotics Laboratory, Chemical Biology Department, Advanced Science Institute, RIKEN 2-1 Hirosawa, Wako, Saitama 351-0198, Japan

Figure 1. The structure of $(-)$ -epicatechin (1) and $(+)$ -catechin (2) .

 \hat{X} This report is Part 12 in the series 'Synthetic studies of proanthocyanidins'. See Ref. [5–15.](#page-6-0)

^{*} Corresponding authors. Tel.: $+81$ 48 467 4839; fax: $+81$ 48 462 1353 (A.S.); tel.: $+81$ 766 56 7500; fax: $+81$ 766 56 2498 (N.N.).

^{0040-4020/\$ –} see front matter © 2009 Elsevier Ltd. All rights reserved. doi:10.1016/j.tet.2009.07.018

effect against a variety of cancers, such as that of the lung, prostate, and breast.^{[16](#page-6-0)} Furthermore, the receptor that mediates the anticancer activity of EGCG was identified, 17 but the effect of the galloyl group in proanthocyanidins is poorly understood.

Our study aimed at synthesizing various pure proanthocyanidins and measuring activity systematically. We previously reported the stereoselective synthesis of procyanidin dimers $(B1,9)$ $(B1,9)$ $(B1,9)$ $B2,9$ $B2,9$ B3,^{5,6} B4,⁹ cis-diastereomer of B3^{[7](#page-6-0)} and their galloyl derivatives¹³⁻¹⁵) with intermolecular condensation^{[5,6,8,13–15](#page-6-0)} and intramolecular^{7,8,11} condensation using TMSOTf as the catalyst. Using this same condensation method, we synthesized procyanidin trimers stereoselectively in good yields[.10,12](#page-6-0) We also reported their antioxidant activities, the Maillard reaction inhibitory activity and DNA polymerase inhibitory activities. Here, we describe the stereo, regio, and oligomerization degree controlled synthesis of two epicatechin series tetramers, epicatechin-(4b-8)-epicatechin-(4b-8)-epicatechin- (4β -8)-epicatechin (3) and epicatechin-(4β -8)-epicatechin-(4β -8)epicatechin-(4b-8)-catechin (arecatannin A1) (4), a epicatechin pentamer, epicatechin-(4b-8)-epicatechin-(4b-8)-epicatechin-(4b-8)-epicatechin-(4 β -8)-epicatechin (5) and two galloyl-substituted

Figure 2. The structure of epicatechin series procyanidin oligomers.

epicatechin series trimers, epicatechin-(4b-8)-epicatechin-(4b-8) epicatechin, $3,3'',3''''$ -O-trigallate (6) and epicatechin-(4 β -8)-epicatechin-(4 β -8)-catechin 3,3",3""-O-trigallate (7) (Fig. 2). Their antioxidant and DNA polymerase inhibitory activities were also described.

2. Result and discussion

2.1. Synthesis

There have been numerous studies concerning isolation, semisynthesis, and bioactivity.¹⁸ Many condensation methods for procyanidin oligomers using Lewis acids were studied after Kawamoto's report, 19 however, systematic study of proanthocyanidins is still difficult even now. As previously mentioned separating purely individual structural analogues from the plant is very difficult because these compounds are presented as mixtures of a number of structurally related compounds. Furthermore, the structural determination of these compounds is not easy. Therefore, a technology for synthesizing these compounds purely and systematically is very useful and important. We already developed and reported on a synthesis methodology applicable to various (4–8) linked procyanidin dimers, trimers, and their gallate derivatives. We report here further study of the synthesis of epicatechin series tetramers (3, 4), pentamers (5), and gallate trimers (6, 7).

The key step is the coupling reaction between the oligomer nucleophile and monomer electrophile using Lewis acid as an activator. We created 2-ethoxyethoxy derivatives on the C-4 position as building blocks for procyanidin synthesis. Treatment of 5,7,3',4'tetrabenzylepicatechin with DDQ in the presence of 2-ethoxyethanol provided the 2-ethoxyethoxy derivative 8.5 8.5 As shown in Scheme 1, condensation of monomer electrophile 8 with trimer nucleophiles 9^{14} 9^{14} 9^{14} and 10^{14} in the presence of TMSOTf in CH₂Cl₂ at -40 °C gave corresponding tetramers 11 and 12 in 88% and 89% yields, respectively. Pentamer 13 was also obtained from the condensation of 8 and tetramer 11 in 84% using same procedure. In the condensation reaction, nucleophiles 9–11 were used in fourfold excess to avoid higher oligomer formation. The 8 reacted with 9–11

Scheme 1. Synthesis of epicatechin tetramers and pentamer. Reagents: (a) TMSOTf, CH₂Cl₂, $-40\,^{\circ}$ C; (b) Pd(OH)₂/C, H₂, THF/MeOH/H₂O; (c) Ac₂O, DMAP, py.

Scheme 2. Synthesis of galloyl epicatechin trimers. Reagents: (a) TMSOTf, CH₂Cl₂, $-40\degree$ C; (b) Pd(OH)₂/C, H₂, THF/MeOH/H₂O.

at C-4 position stereoselectively to give 3,4-trans oligomers preferentially. These protected oligomers were hydrogenated in the presence of $Pd(OH)₂/C$ and purified with LH-20 short column and HPLC to give free tetramers 3, 4, and pentamer 5 in 74%, 58%, and 68% yields, respectively. The all spectral data of synthesized 3, 5, and peracetate of 3 (14) agreed with the those of the reported data.^{3a}

The synthesis of two $3,3'',3''''$ -O-trigallates substituted trimers is shown in Scheme 2. Electrophile 15^{14} 15^{14} 15^{14} was condensed with nucle-ophiles 16 and 17^{[14](#page-6-0)} in the presence of TMSOTf at -20 °C to afford trimers 18 and 19 with yields of 59% and 76%, respectively. And the deprotection of all benzyl groups under hydrogenation conditions and purification yielded tetramers 6 and 7 in yields of 83% and 54%, respectively.^{[20](#page-6-0)}

2.2. Biological activity of synthetic procyanidin oligomers

2.2.1. DPPH radical scavenging activity 21

Proanthocyanidins are known as strong antioxidants and radical scavengers.^{[22](#page-6-0)} In previous papers,¹²⁻¹⁴ we investigated the DPPH radical scavenging activity of synthesized procyanidin oligomers, and 3-O-gallate dimers.

The SC_{50} values (the concentration of 50% scavenging activity) of monomeric (–)-epicatechin (1), (+)-catechin (2), and synthesized trimers to pentamer (3–7) were 2.2, 2.6, 0.4, 0.4, 0.4, 0.4, and 0.6μ M, respectively. It appears that these oligomers have significant radical scavenging activity, compared to monomeric compounds and DL- α -Tocopherol. It was shown that elongation of oligomer length and modification of 3-O-positions with galloyl groups are effective for DPPH radical scavenging activity, compared to trimers epicatechin-(4b-8)-epicatechin-(4b-8)-epicatechin (20)^{[10](#page-6-0)} and epicatechin-(4β-8)-epicatechin-(4β-8)-catechin (21)¹⁰ (Table 1, Fig. 3).

2.3. Mammalian DNA polymerase inhibitory activities

Monomeric flavan-3-O-gallates, (–)-epicatechin-3-O-gallate, (-)-epigallocatechin-3-O-gallate, etc., are known as inhibitors of DNA and RNA polymerases, 23 and it was apparent that the galloyl group appears essential for the inhibitory effect. As we described in a previous paper, 3-O-galloyl-substituted flavan-3-ols actively inhibit against DNA polymerase $\alpha^{.13-15,24}$ DNA polymerase α , which is

Table 1 DPPH radical scavenging activity

| Entry | Compound | SC_{50} values (μ M) | | |
|-------|-----------------|-----------------------------|--|--|
| | | 2.2 | | |
| า | 2 | 2.6 | | |
| 3 | 3 | 0.4 | | |
| | 4 | 0.4 | | |
| 5 | 5 | 0.4 | | |
| 6 | 6 | 0.4 | | |
| | 7 | 0.6 | | |
| 8 | 20 | 0.7 | | |
| 9 | 21 | 0.7 | | |
| 10 | DL-α-Tocopherol | 17 | | |

Figure 3. Structure of procyanidin trimers.

a DNA replicative polymerase, is regarded as the target of some anticancer drugs, because DNA polymerase plays a central roles in the DNA replication indispensable for the prolification of cancer cells. Thus, we expect galloyl-substituted procyanidin oligomers to be effective inhibitors of DNA polymerase. As our previous report demonstrated that procyanidin dimers have moderate DNA poly m erase inhibitory activity, whereas monomeric $(-)$ -epicatechin and $(+)$ -catechin have none.¹³

We also interested in the relationships between activity and oligomer length. [Table 2](#page-3-0) shows the IC_{50} values of synthesized procyanidin oligomers against calf DNA polymerase α and rat DNA polymerase b. All oligomeric procyanidins inhibit DNA polymerase α , and the activity tends to be stronger with the oligomer length.(-)-Epicatechin pentamer 5, in particular, potently inhibited

Table 2 IC_{50} values of compounds on the activities of mammalian DNA polymerases α and B^2

| Entry | Compound | IC_{50} values (μ M) | | |
|-------|----------|------------------------------|----------------------------|--|
| | | Calf DNA polymerase α | Rat DNA polymerase β | |
| | | >100 | >100 | |
| | | >100 | >100 | |
| | 3 | $0.178 + 0.009$ | 31.2 ± 1.6 | |
| | 5 | 0.075 ± 0.004 | 26.0 ± 1.3 | |
| 5 | 6 | 0.230 ± 0.011 | 38.5 ± 1.9 | |
| 6 | | 0.243 ± 0.012 | 40.7 ± 2.1 | |
| | 20 | 0.575 ± 0.028 | $57.9 + 3.0$ | |

Calf DNA polymerase α , and the substitution of 3-O-position with galloyl groups was highly efficacious in inhibition of DNA polymerase a.

3. Conclusion

We have developed a simple and versatile method of synthesizing procyanidin oligomers. We showed here the synthesis of (-)-epicatechin series tetramers, pentamer, and 3-O-galloyl trimers. DPPH radical scavenging activity, and DNA polymerase inhibitory activity of synthesized oligomers were also investigated. All of the oligomers showed strong DPPH radical scavenging activity. Furthermore, synthesized oligomers had potent inhibitory activity against DNA polymerase α . The elongation of oligomers and the substitution of 3-O-position with the galloyl group were shown to be highly effective for DNA polymerase inhibition.

4. Experimental

4.1. Synthesis

Optical rotation was measured with a Horiba SEPA-300 spectrometer. ¹H NMR spectra were measured with JEOL JNMLA400 spectrometer. MS spectra were recorded with a JEOL JMS-AX500 instrument. HPLC purification was carried out on a Mightysil[®] RP-18 GP column (Kanto Chemical Co. Inc, Japan; 250×20 mm, 5 mm) using the solvents (A) 0.05% CF₃CO₂H in CH₃CN and (B) 0.05% $CF₃CO₂H$ in H₂O. Elution was done with a linear gradient 5-100% A in 40 min (flow rate, 4.0 mL/min).

4.1.1. [4,8:4",8":4"",8""]-2,3-cis-3,4-trans:2",3"-cis-3",4"trans:2"",3""-cis-3"",4""-trans-2""",3"""-cis-Hexadeca-O-benzyl-tetra-(–)-epicatechin (**11**)

To a solution of 9 (779 mg, 0.40 mmol) and 8 (73.9 mg, 0.10 mmol) in CH_2Cl_2 (80 mL) was added dropwise TMSOTf (0.20 mL, 0.10 mmol, 0.5 M solution in CH_2Cl_2) at -40 °C. After stirring for 5 min, the pale yellow reaction mixture was quenched with satd sodium hydrogen carbonate. The aq solution was extracted with CHCl₃ and the organic phase was washed with water and brine, and dried ($Na₂SO₄$). Filtration, concentration, and preparative silica gel TLC purification (hexane/EtOAc, 2/1) afforded a 228 mg of 11 (0.088 mmol, 88%) of as a colorless oil: $[\alpha]_D^{24}$ +104.8 (c 0.94, CHCl₃), $[\alpha]_D^{24}$ +80.5 (c 1.00, EtOAc), {lit.^{[3a](#page-6-0)} $[\alpha]_D$ $+85.9$ (c 12.0 g L⁻¹, EtOAc)}; ¹H NMR (400 MHz, CDCl₃, 0.5:0.5 mixture of rotational isomers) 7.50–6.39 (89H, m), 6.39 (0.5H, s), 6.37 (0.5H, d, J=8.3 Hz), 6.33 (0.5H, d, J=8.3 Hz), 6.27 (0.5H, d, $J=2.2$ Hz), 6.23 (0.5H, s), 6.14 (0.5H, dd, $J=1.7$, 8.3 Hz), 6.07 (0.5H, d, J = 2.2 Hz), 6.05 (0.5H, d, J = 8.3 Hz), 6.03 (0.5H, dd, J = 1.7, 8.3 Hz), 5.91 (0.5H, s), 5.90 (0.5H, s), 5.88 (0.5H, s), 5.86 (0.5H, d, $J=2.2$ Hz), 5.75 (0.5H, s), 5.75 (0.5H, br s), 5.63 (0.5H, d, $J=2.2$ Hz), 5.23 (0.5H, br s), 5.43 (0.5H, dd, $J=1.7$, 8.3 Hz), 5.39 (0.5H, br s), 5.37 (0.5H, br s), 5.23–3.92 (37H, m), 4.30–4.22 (0.5H, m), 4.25– 4.20 (0.5H, m), 4.11–4.05 (1H, m), 4.01–3.93 (1.5H, m), 3.68–3.64

 $(0.5H, m)$, 2.93-2.83 (2H, m), 1.80 (0.5H, d, J=5.8 Hz, OH), 1.60 (0.5H, d, J=6.4 Hz, OH), 1.53-1.52 (0.5H, m, OH), 1.43 (0.5H, d, $J=7.3$ Hz, OH), 1.37 (0.5H, d, $J=7.5$ Hz, OH), 1.35 (0.5H, d, $J=7.0$ Hz, OH), 1.15 (0.5H, d, J=7.3 Hz, OH), 1.06 (0.5H, d, J=8.0 Hz, OH); ¹³C NMR (100 MHz, CDCl3) 158.3, 158.05, 158.03, 157.3, 156.7, 156.6, 156.4, 156.3, 156.2, 156.14 (3), 156.06, 156.00, 155.70, 155.67, 155.4, 155.2, 154.8, 153.2, 153.1, 152.80 (2), 152.76, 149.3, 149.1, 149.0, 148.91, 148.87, 148.86, 148.83, 148.77, 148.6, 148.4 (2), 148.3, 148.14, 148.08, 147.8, 147.7, 138.8, 138.1, 137.6-136.9 (C×32), 133.0, 132.57, 132.54, 132.52 (2), 132.3, 131.3 (2), 128.6–125.9 $(C \times 94)$, 119.9, 119.7, 119.2, 119.0 (\times 2), 118.72, 118.67, 118.5, 115.1 $(x2)$, 115.00, 114.95, 114.82, 114.77, 114.1, 113.6 $(x3)$, 113.4, 113.21, 113.15 (\times 2), 112.8, 111.2, 110.99, 110.96, 110.6, 110.2, 110.0, 106.3, 105.6, 105.1, 104.8, 104.5, 104.2, 101.2, 101.0, 94.4, 93.9 (2), 93.4, 92.5, 92.4, 92.3, 92.2, 91.3, 90.4, 78.1, 78.0, 76.2, 75.85 (\times 2), 75.74, 75.66 $(\times 2)$, 72.8, 72.6, 72.5, 71.7, 71.4–69.1 $(C \times 34)$, 64.9, 60.4, 36.9, 36.3, 35.6, 35.5, 35.1 (\times 2), 28.5 (\times 2); IR (neat, cm⁻¹) 3569 (m), 3400 (br), 3065 (m), 3030 (m), 2930 (m), 2361 (m), 2342 (m), 1952 (w), 1850 (w), 1813 (w), 1599 (s), 1512 (s), 1454 (s), 1423 (m), 1383 (m), 1265 (m), 1217 (m), 1121 (s), 1076 (m), 1026 (m), 910 (w), 852 (w), 752 (m); FABMS (m/z) 2620 (60), 2619 (95), 2618 $([M+Na]^+, 21)$, 2598 (27), 2597 (51), 2596 $([M+H]^+, 19)$, 2438 (60), 2355 (72), 2156 (88), 2068 (100), 1770 (63), 1640 (62), 1322 (62), 1112 (79); FAB-HRMS calcd for $C_{172}H_{147}O_{24}$ [M+H]⁺, 2596.0282; found: 2596.0212.

4.1.2. [4,8:4",8":4"",8""]-2,3-cis-3,4-trans:2",3"-cis-3",4"trans:2"",3""-cis-3"",4""-trans-2""",3"""-trans-Hexadeca-Obenzyl-(-)-epicatechin-(-)-epicatechin-(-)-epicatechin- $(+)$ -catechin (12)

To a solution of **10** (277 mg, 0.14 mmol) and **8** (26.2 mg, 0.036 mmol) in $CH₂Cl₂$ (50 mL) was added dropwise TMSOTf (0.072 mL, 0.036 mmol, 0.5 M solution in CH_2Cl_2) at -40 °C. After stirring for 5 min, the pale yellow reaction mixture was quenched with satd sodium hydrogen carbonate. The aq solution was extracted with $CHCl₃$ and the organic phase was washed with water and brine, and dried ($Na₂SO₄$). Filtration, concentration, and preparative silica gel TLC purification (hexane/EtOAc, 2:1) afforded a 83 mg of **12** (0.032 mmol, 89%) of as a colorless oil: [α] $_{D}^{24}$ +102.4 (α 0.76, CHCl₃); ¹H NMR (400 MHz, CDCl₃, 0.5:0.5 mixture of rotational isomers) 7.53–5.89 (89.5H, m), 6.39 (0.5H, s), 6.33 (0.5H, d, J=8.3 Hz), 6.26 (0.5H, d, J=2.2 Hz), 6.22 (0.5H, s), 6.17 (0.5H, dd, $J=1.7$, 8.3 Hz), 6.07 (0.5H, d, J=2.2 Hz), 6.04 (0.5H, d, J=8.3 Hz), 6.01 $(0.5H, dd, J=1.7, 8.3 Hz)$, 5.92 $(0.5H, s)$, 5.91 $(0.5H, s)$, 5.89 $(0.5H, s)$, 5.85 (0.5H, d, J=2.2 Hz), 5.74 (0.5H, s), 5.74 (0.5H, br s), 5.63 (0.5H, d, $J=2.2$ Hz), 5.52 (0.5H, br s), 5.41–5.36 (0.5H, m), 5.39 (0.5H, br s), 5.37 (0.5H, br s), 5.22 (0.5H, br s), 5.20–4.32 (34.5H, m), 5.17 (0.5H, d, J=1.9 Hz), 5.14-5.13 (0.5H, m), 4.22 (0.5H, d, J=14.1 Hz), 4.14 $(0.5H, d, J=12.2 Hz)$, 4.09-4.07 $(0.5H, m)$, 4.00-3.97 (1H, m), 3.97-3.93 (0.5H, m), 3.91–3.88 (0.5H, m), 3.65 (0.5H, br s), 3.47–3.38 $(0.5H, m)$, 3.30–3.25 $(0.5H, m)$, 2.99 $(0.5H, dd, J=5.6, 16.3 Hz)$, 2.92 $(0.5H, dd, J=5.4, 16.1 Hz)$, 2.60 $(0.5H, dd, J=9.3, 16.1 Hz)$, 2.54 $(0.5H,$ dd, J=9.3, 16.3 Hz), 1.79 (0.5H, d, J=5.6 Hz, OH), 1.58 (0.5H, d, $J=6.3$ Hz, OH), 1.55–1.50 (1H, m, OH), 1.48 (0.5H, d, $J=3.1$ Hz, OH), 1.41 (0.5H, d, J=7.1 Hz, OH), 1.22 (0.5H, d, J=6.1 Hz, OH), 1.18 (0.5H, d, J=7.1 Hz, OH); ¹³C NMR (100 MHz, CDCl₃) 158.3, 158.0 (\times 2), 157.3, 156.7, 156.6, 156.3, 156.2 (\times 2), 156.0 (\times 3), 155.68, 155.65, 155.45, 155.40, 155.35, 155.2, 154.7, 153.22, 153.18, 152.85 (2), 152.81, 149.2 (\times 2), 149.00, 148.95 (\times 2), 148.91, 148.83 (\times 2), 148.77, 148.76, 148.5, 148.2, 148.03, 147.96, 147.8, 147.6, 138.8, 138.1, 137.6–136.8 $(C \times 32)$, 132.9, 132.57 $(\times 2)$, 132.55 $(\times 2)$, 132.3, 131.4, 131.3, 128.6– 125.9 (C \times 94), 120.1 (\times 2), 119.9, 119.7, 119.2, 119.0 (\times 2), 118.5, 115.1– 112.8 (C×16), 111.1, 110.8, 110.7, 110.4, 110.1, 109.9, 106.3, 105.6, 105.3, 104.8, 104.5, 104.4, 102.2, 101.9, 94.4, 93.9, 93.8, 93.4, 92.5, 92.1, 91.9, 91.8, 90.9, 90.4, 81.2 (\times 2), 76.1, 75.8, 75.6 (\times 2), 75.5 (\times 2), 72.74, 72.72, 72.55, 72.50, 71.9–69.0 (C34), 68.10, 68.08, 36.8, 36.3, 35.7,

35.4, 35.1, 35.0, 27.4 (\times 2); IR (neat, cm $^{-1}$) 3569 (m), 3450 (br), 3088 (w), 3065 (m), 3080 (m), 2930 (m), 2872 (m), 1954 (w), 1869 (w), 1811 (w), 1601 (s), 1510 (s), 1423 (s), 1327 (m), 1265 (s), 1217 (s), 1122 (s), 1026 (s), 910 (w), 854 (w), 754 (s); FABMS (m/z) 2618 ([M+Na]⁺, 46), 2598 (30), 2596 ([M+H]⁺, 38), 2524 (45), 2440 (43), 2163 (72), 2134 (49), 1824 (100).

4.1.3. [4,8:4",8":4"",8""]-2,3-cis-3,4-trans:2",3"-cis-3",4"-trans:2"", 3""-cis-3"",4""-trans-2""",3"""-cis-3""",4"""-trans:2"" "" 3"" ""-cis-Eicosa-O-benzyl-(-)-epicatechin-(-)-epicatechin-(–)-epicatechin-(–)-epicatechin-(–)-epicatechin (**13**)

To a solution of 11 (260 mg, 0.10 mmol) and 8 (19.5 mg, 0.025 mmol) in CH_2Cl_2 (40 mL) was added dropwise TMSOTf (0.050 mL, 0.025 mmol, 0.5 M solution in CH_2Cl_2) at -40 °C. After stirring for 5 min, the pale yellow reaction mixture was quenched with satd sodium hydrogen carbonate. The aq solution was extracted with CHCl $_3$ and the organic phase was washed with water and brine, and dried ($Na₂SO₄$). Filtration, concentration, and preparative silica gel TLC purification (hexane/EtOAc, 2:1, twice development) afforded a 67 mg of 13 (0.021 mmol, 84%) of as a colorless oil: [α] $_{{\rm D}}^{23}$ +107.2 (c 0.43, CHCl3), [α] $_{{\rm D}}^{25}$ +102.1 (c 1.26, EtOAc) {lit. 3a 3a 3a [α] $_{\rm D}$ +100 (c 13.0 g L $^{-1}$, EtOAc)}; 1 H NMR (400 MHz, CDCl3, 0.5:0.5 mixture of rotational isomers) 7.53–5.39 (115.5H, m), 6.39 (0.5H, s), 6.23 (0.5H, s), 6.04 (0.5H, d, J=2.2 Hz), 5.95 (0.5H, s), 5.93 (0.5H, s), 5.88-5.87 (1H, s), 5.86 (0.5H, d, J=2.2 Hz), 5.75 (0.5H, br s), 5.71 (0.5H, s), 5.62 (0.5H, d, J=2.2 Hz), 5.50 (0.5H, br s), 5.47 (0.5H, d, J=2.2 Hz), 5.43 (0.5H, br s), 5.39 (0.5H, br s), 5.26 (0.5H, br s), 5.25–4.15 (47.5H, m), 4.27 (1H, br s), 4.10 (0.5H, br), 4.04 (0.5H, br), 4.01–3.93 (2.0H, m), 3.88 (0.5H, d, $J=7.1$ Hz), 3.66 (0.5H, br), $3.00-2.85$ (2H, m), 1.80 (0.5H, d, $J=6.1$ Hz, OH), 1.60 (0.5H, d, J=5.9 Hz, OH), 1.42 (0.5H, br, OH), 1.42-1.40 (0.5H, m, OH), 1.38 (0.5H, d, J=7.3 Hz, OH), 1.32 (0.5H, d, J=7.4 Hz, OH), 1.31 (0.5H, d, J=7.0 Hz, OH), 1.26-1.21 (0.5H, m, OH), 1.14 (0.5H, d, J=7.8 Hz, OH), 1.10 (0.5H, d, J=7.6 Hz, OH); ¹³C NMR (100 MHz, CDCl₃) 158.3, 158.05 (\times 2), 158.02, 157.3, 156.7, 156.6, 156.41, 156.37, 156.29 (\times 3), 156.17 (\times 2), 156.0 (\times 3), 156.9 (\times 2), 155.5, 155.4, 155.2, 154.8, 153.2 (2), 153.1, 153.94, 152.95, 152.8, 152.7, 149.2–147.6 $(C \times 20)$, 138.8, 138.1, 138.0, 137.9, 137.7–136.8 $(C \times 40)$, 133.0, 132.62 $(x2)$, 132.57 $(x2)$, 132.50 $(x2)$, 132.3, 131.2 $(x2)$, 128.8–126.0 (C×136), 119.9, 119.8, 119.3, 118.99, 118.95, 118.8, 118.7, 118.6, 118.5 $(x2)$, 115.1–112.8 (C \times 20), 111.19, 111.16, 111.0, 110.6 (\times 2), 110.2, 110.1, 110.0, 106.4, 105.7, 105.4, 105.0, 104.8, 104.6, 104.5 (2), 101.2, 101.1, 94.3, 93.8 (\times 3), 93.4, 92.5, 92.2 (\times 2), 97.8 (\times 2), 91.4, 90.4, 78.0, 77.9, 76.1, 75.94, 75.86 (\times 2), 75.80 (\times 2), 75.66 (\times 2), 72.7, 72.6 (\times 2), 71.7 (\times 2), 71.4–69.1 (C \times 27), 64.9, 64.8, 37.0 (\times 2), 36.3, 35.7, 35.4, 35.3, 35.1, 34.0, 28.5 $(\times 2)$; IR (neat, cm $^{-1}$) 3569 (m), 3065 (m), 3030 (m), 2932 (m), 2874 (m), 1954 (w), 1869 (w), 1811 (w), 1734 (w), 1597 (s), 1514 (s), 1423 (s), 1375 (s), 1266 (s), 1217 (s), 1120 (s), 1026 (s), 910 (w), 852 (w), 750 (m); FABMS (m/z) 3271 (25), 3270 (34), 3269 (39), 3268 (81), 3267 (40), 3266 $([M+Na]^+, 39)$, 3246 (70), 3246 (91), 3245 (87), 3244 ($[M+H]^+,$ 20), 2821 (56), 2806 (41), 2805 (46), 2694 (46), 2518 (48), 2486 (58), 1043 (100).

4.1.4. [4,8:4",8":4"",8""]-2,3-cis-3,4-trans:2",3"-cis-3",4"trans:2"",3""-cis-3"",4""-trans-2""",3"""-cis-Tetra- $(-)$ -epicatechin (3)

A solution of 11 (83 mg, 0.032 mmol) in 22 mL of THF/MeOH/ H₂O, 20:1:1 was hydrogenated over 20% Pd(OH)₂/C (5 mg) for 8 h at rt. Filtration and concentration afforded a pale brown solid, which was purified by Sephadex $^{\circledast}$ LH-20 short column chromatography (MeOH) and HPLC purification to give 27.4 mg of pure 3 (0.025 mmol, 74%) as an amorphous solid: [α] $_{{\rm D}}^{25}$ +102.9 (c 0.11, MeOH) {lit. 3a 3a 3a [$\alpha]_{\rm D}$ +93.3 (c 9.3 g L $^{-1}$, MeOH)}; 1 H NMR (400 MHz, CD₃OD, -20 °C) 7.13 (1H, d, J=1.7 Hz), 7.10 (1H, d, J=1.7 Hz), 7.03 $(1H, d, J=1.7 Hz)$, 6.91 $(1H, dd, J=1.7, 8.3 Hz)$, 6.90 $(1H, d, J=1.7 Hz)$,

6.79–6.68 (7H, m), 6.01 (1H, d, J=2.2 Hz), 5.98 (1H, d, J=2.2 Hz), 5.95 (1H, s), 5.94 (1H, s), 5.93 (1H, s), 5.31 (1H, d, J=2.2 Hz), 5.30– 5.20 (1H, m), 5.09 (1H, br s), 5.10–4.95 (1H, m), 4.75 (1H, br s), 4.73– 4.71 (2H, br s), 4.35–4.31 (1H, m), 4.07 (1H, d, J=1.7 Hz), 3.99 (1H, d, J=2.6 Hz), 3.97 (1H, d, J=2.0 Hz), 2.95 (1H, dd, J=2.7, 16.4 Hz), 2.81 (1H, d, J=16.4 Hz); ¹³C NMR (100 MHz, CD₃OD, -20 °C) 158.1, 157.9, 157.8, 157.0, 156.9, 156.7, 156.6, 156.3 (\times 2), 155.0, 154.8, 154.6, 146.0, 145.9, 145.7, 145.7, 145.6, 145.40, 145.38, 145.2, 132.64, 132.56, 132.51, 132.0, 119.1, 118.8, 118.6 (\times 2), 116.0, 115.8 (\times 2), 115.8, 115.1, 115.0 (2), 114.8, 107.6, 107.3, 107.1, 102.4, 102.2, 101.8, 100.0, 97.3, 97.0, 96.8, 95.8, 95.7, 79.5, 76.9, 76.8, 76.7, 73.5, 73.1, 72.9, 66.7, 37.4, 37.3, 37.1, 30.0; FABMS (m/z) 1179 (11), 1178 (12), 1177 ($[M+Na]$ ⁺, 30), 1176 (17), 1156 (14), 1155 ([M+H]⁺, 24), 1154 (21), 1153 (24), 999 (20), 867 (27), 866 (29), 578 (100); FAB-HRMS calcd for $C_{60}H_{51}O_{24}$ $[M+H]$ ⁺, 1155.2770; found: 1155.2822.

4.1.5. [4,8:4",8":4"",8""]-2,3-cis-3,4-trans:2",3"-cis-3",4"trans:2"",3""-cis-3"",4""-trans-2""",3"""-trans-(-)-Epicatechin-(–)-epicatechin-(–)-epicatechin-(+)-catechin (arecatannin A1) (**4**)

A solution of 12 (66 mg, 0.025 mmol) in 22 mL of THF/MeOH/ H₂O, 20:1:1 was hydrogenated over 20% Pd(OH)₂/C (5 mg) for 12 h at rt. Filtration and concentration afforded a pale brown solid, which was purified by Sephadex® LH-20 short column chromatography (MeOH) and HPLC purification to give 16.8 mg of pure 4 (0.015 mmol, 58%) as an amorphous solid. $[\alpha]_D^{25}$ +73.3 (c 0.10, MeOH), lit.^{[28](#page-6-0)} {[α]₅₈₉ +100}; ¹H NMR (400 MHz, CD₃OD, -20 °C, a small mount of rotational isomer was observed) 7.10 (1H, d, $J=1.7$ Hz), 7.05 (1H, d, $J=1.7$ Hz), 7.13–6.68 (10H, m), 6.02–5.88 (5H, m), 5.33 (1H, br s), 5.26–5.19 (1H, m), 5.08 (1H, br s), 5.12 (1H, d, J=4.9 Hz), 4.78 (1H, br s), 4.75 (1H, br s), 4.72 (1H, br s), 4.19 (1H, ddd, $J=4.6$, 4.8, 4.9 Hz), 4.11 (1H, br s), 4.10 (1H, br s), 3.97 (1H, br s), 2.62 (1H, dd, J=4.6, 16.8 Hz), 2.55 (1H, dd, J=4.8, 16.8 Hz); ¹³C NMR $(100 \text{ MHz}, \text{CD}_3 \text{OD}, -20 \text{ }^{\circ}\text{C})$ 157.9, 157.8, 157.11, 157.06, 156.8, 156.4 $(x2)$, 155.8, 154.9, 154.83, 154.80, 153.7, 146.1, 145.84, 145.78, 145.74, 145.68, 145.4, 145.3, 145.2, 132.7, 132.62, 132.57, 132.4, 119.2, 119.1, 118.7, 118.6, 116.01 $(\times 2)$, 115.98, 115.95, 115.8, 115.0 $(\times 3)$, 114.9, 114.8, 113.9, 108.2, 107.2, 106.8, 102.7, 102.1, 101.8, 100.2, 97.04, 97.02, 81.6, 76.94, 76.85, 76.7, 73.5, 73.0, 72.1, 68.2, 37.4, 37.2, 37.1, 26.3; FABMS $(m|z)$ 1178 (19), 1177 ($[M+Na]$ ⁺, 31), 1176 (15), 1175 (11), $1156(13)$, $1155([M+H]^{+}$, 83), 1031 (43), 973 (39), 957 (38), 937 (43), 846 (43), 666 (100); FAB-HRMS calcd for $C_{60}H_{50}O_{24}$ [M+Na]⁺, 1177.2590; found: 1177.2629.

```
4.1.6. [4,8:4",8":4"",8""]-2,3-cis-3,4-trans:2",3"-cis-3",4"-
trans:2"",3""-cis-3"",4""-trans-2""",3"""-cis-3""",4"""-
trans:2"" "",3"" "" -cis-Penta-(-)-epicatechin (5)
```
A solution of 13 (30 mg, 0.0092 mmol) in 22 mL of THF/MeOH/ H₂O, 20:1:1 was hydrogenated over 20% Pd(OH)₂/C (5 mg) for 4 h at rt. Filtration and concentration afforded a pale brown solid, which was purified by Sephadex $^{\circledast}$ LH-20 short column chromatography (MeOH) and HPLC purification to give 9.0 mg of pure 5 (0.0062 mmol, 68%) as an amorphous solid. $[\alpha]_D^{24}$ +96.0 (c 0.17, MeOH) { $\rm{lit.}^{3a}$ [α]_D +116 (c 8.3 g L⁻¹, MeOH)}; ¹H NMR (400 MHz, CD₃OD, -20 °C) 7.13 (1H, d, J=1.7 Hz), 7.11 (1H, d, J=1.7 Hz), 7.10 $(1H, d, J=1.7 Hz)$, 7.04 $(1H, d, J=1.7 Hz)$, 6.92 $(1H, d, J=1.7 Hz)$, 6.91 $(1H, dd, J=1.7, 8.3 Hz)$, 6.80 $(1H, dd, J=1.7, 8.3 Hz)$, 6.80–6.69 (8H, m), 6.02 (1H, d, J=2.2 Hz), 5.98 (1H, d, J=2.2 Hz), 5.98 (1H, s), 5.95 (1H, s), 5.93 (2H, s), 5.31 (1H, br s), 5.30 (1H, br s), 5.27 (1H, br s), 5.09 (1H, br s), 5.00 (1H, br s), 4.78 (1H, br s), 4.77 (1H, br s), 4.73 $(1H, br s)$, 4.71 $(1H, br s)$, 4.35-4.30 $(1H, m)$, 4.11 $(1H, d, J=2.2 Hz)$, 4.10 (1H, d, J=2.0 Hz), 4.00 (1H, d, J=1.7 Hz), 3.98 (1H, d, J=2.2 Hz), 2.96 (1H, dd, J=3.4, 16.1 Hz), 2.81 (1H, d, J=16.1 Hz); ¹³C NMR (100 MHz, CD₃OD, -20 °C) 158.2, 158.0, 157.9, 157.1, 157.0, 156.9, 156.7, 156.6, 156.4, 156.3, 156.2, 155.1, 155.0, 154.9, 154.7, 146.0, 145.9, 145.8, 145.74, 145.69, 145.66, 145.41, 145.40, 145.3, 145.2, 132.63 (×2), 132.58, 132.54, 132.0, 119.0, 118.9, 118.73, 118.65 (×2),

116.03, 115.97, 115.86 (×2), 115.78 (×2), 115.1, 115.0 (×2), 114.9, 107.6, 107.5, 107.4, 107.2, 102.5, 102.4, 102.3, 101.8, 100.0, 97.3 (\times 2), 97.2, 97.0, 96.2, 96.0, 79.5, 76.9, 76.80, 76.76, 76.6, 73.5, 73.1, 72.9 $(x2)$, 66.7, 37.5 $(x2)$, 37.3, 37.1, 30.0; FABMS (m/z) 1446 (25), 1465 $([M+Na]^+, 33)$, 1464 (21), 1444 (9.6), 1443 $([M+H]^+, 34)$, 1442 (20), 1177 (28), 1162 (31), 1100 (37), 1024 (29), 981 (33), 944 (31), 886 (32), 746 (50), 694 (69), 693 (100); FAB-HRMS calcd for $C_{75}H_{63}O_{30}$ $[M+H]$ ⁺, 1443.3404; found: 1443.3331.

4.1.7. Peracetate of 3 (14)

Acetylation of 3 (5.9 mg, 5.0 μ mol) with general procedure gave **14** (9.5 mg, 4.8 µmol, 95%) as colorless amorphous. α ²⁵ +68.1 (c 0.45, CHCl₃), [α] $_{\rm D}^{24}$ +79.9 (c 0.48, EtOAc) {lit.^{[3a](#page-6-0)} [α]_D +82.2 (c 18.7 g L $^{-1}$, EtOAc)}; ¹H NMR (400 MHz, CDCl₃, 0.5:0.5 mixture of rotational isomers) 7.36–6.69 (12H, m), 6.89 (0.5H, s, 6), 6.77 $(0.5H, d, J=2.2 Hz, AB), 6.75 (0.5H, s, 6), 6.71 (0.5H, s, 6), 6.65$ $(0.5H, d, J=2.2 Hz, A6), 6.64 (0.5H, s, 6), 6.62 (0.5H, s, 6), 6.58$ (0.5H, s, 6), 6.25 (0.5H, d, J=2.2 Hz, A8), 5.88 (0.5H, d, J=2.2 Hz, A6), 5.74 (0.5H, br s), 5.50–5.46 (1H, m, $L3\times2$), 5.45 (1H, br s), 5.36 (1H, br s), 5.34 (0.5H, br s), 5.31 (0.5H, br s), 5.29 (1H, br s), 5.22–5.18 (1H, br s, L2 \times 2), 5.15 (0.5H, br s), 4.97–4.94 (0.5H, m), 4.84 (0.5H, br s), 4.80 (0.5H, br s), 4.76 (0.5H, br s), 4.66 (0.5H, br s), 4.61 (0.5H, br s), 4.56 (0.5H, br s), 4.51 (0.5H, d, $J=2.2$ Hz), 3.10–3.05 (1H, m, L4), 2.99–2.93 (1H, m, L4), 2.38 (1.5H, s, Ac), 2.32 (1.5H, s, Ac), 2.30 (1.5H, s, Ac), 2.290 (1.5H, s, Ac), 2.286 (1.5H, s, Ac), 2.281 (1.5H, s, Ac), 2.27 (6H, s, Ac), 2.26 (3H, s, Ac), 2.25 (1.5H, s, Ac), 2.245 (1.5H, s, Ac), 2.239 (1.5H, s, Ac), 2.22 (4.5H, s, Ac), 2.18 (1.5H, s, Ac), 2.17 (1.5H, s, Ac), 2.05 (1.5H, s, Ac), 1.99 (3H, s, Ac), 1.97 (1.5H, s, Ac), 1.94 (1.5H, s, Ac), 1.91 (1.5H, s, Ac), 1.89 (1.5H, s, Ac), 1.88 (3H, s, Ac), 1.86 (1.5H, s, Ac), 1.85 (1.5H, s, Ac), 1.79 (1.5H, s, Ac), 1.77 (1.5H, s, Ac), 1.64 (1.5H, s, Ac), 1.52 (1.5H, s, Ac), 1.45 (1.5H, s, Ac), 1.44 (1.5H, s, Ac), 1.43 (1.5H, s, Ac), 1.35 (1.5H, s, Ac), 1.25 (1.5H, s, Ac); ¹³C NMR (100 MHz, CDCl₃) 170.1 (\times 2), 170.0 (\times 2), 169.2, 169.1, 169.0, 168.9 (\times 2), 168.8, 168.7 $(x3)$, 168.5, 168.39, 168.36, 168.32, 168.29, 168.18, 168.11, 168.1 167.7 (C19), 167.6, 155.9, 154.9, 154.1, 151.91, 151.88, 151.84, 151.7, 151.6, 149.92, 149.88, 149.85, 149.0, 148.7, 148.6, 148.53, 148.48, 148.0, 147.9, 147.5 (\times 2), 147.4, 147.3 (\times 2), 147.2, 142.24, 142.18, 142.16, 142.09, 142.07 (×3), 142.0, 141.94, 141.89, 141.84, 141.81 $(x3)$, 141.73, 141.69, 136.5, 135.8 $(x2)$, 135.5, 135.38, 135.35, 135.1, 133.1, 124.9-121.4 (C×26), 118.5, 118.2, 117.8 (×2), 117.41, 117.39, 112.7, 112.6, 112.3, 111.7, 111.3, 110.8 (\times 2), 110.7, 110.6, 109.9, 109.8, 109.4, 108.2, 107.6, 107.3, 77.6, 77.2-76.6 (C×3), 76.4, 75.4, 75.1 $(x2)$, 74.7, 73.9, 71.5, 71.3, 71.0, 70.7, 66.6 $(x2)$, 35.7, 35.3, 35.2 $(x2)$, 34.3, 32.9, 26.4 $(x2)$, 21.1, 21.0, 20.78, 20.76, 20.7–20.6 $(C \times 18)$, 20.54, 20.46, 20.44, 20.32, 20.31 $(\times 2)$, 20.20, 20.17, 20.12, 20.11, 29.05 (\times 2), 19.99, 19.7, 19.6, 19.5 (\times 2), 19.3; IR (neat, cm $^{-1})$ 3063 (m), 2988 (m), 2936 (m), 2853 (w), 2413 (w), 2309 (w), 2091 (w), 1748 (s), 1600 (s), 1507 (s), 1429 (s), 1372 (s), 1262 (s), 1206 (s), 1109 (s), 1043 (s), 976 (m), 949 (m), 902 (s), 841 (m); FABMS (m/z) 2021 (28), 2020 (59), 2018 (100), 2017 ([M+Na]⁺, 100), 1977 (24), 1975 (47), 1894 (29), 1893 (39), 1892 (36), 1791 (28), 1790 (30), 1789 (28), 1382 (26), 1172 (29), 972 (30), 971 (32); FAB-HRMS calcd for C₁₀₀H₉₀O₄₄Na [M+Na]⁺, 2017.4703; found: 2017.4696.

4.1.8. [4,8:4",8"]-2,3-cis-3,4-trans:2",3"-cis-3",4"-trans:2"",3""cis-Dodeca-O-benzyl-tri-(-)-epicatechin-3,3",3"" -O-(tri-Obenzyl)gallate (18)

To a solution of 16 (284 mg, 0.13 mmol) and 15 (35.0 mg, 0.030 mmol) in CH_2Cl_2 (60 mL) was added dropwise TMSOTf (0.036 mL, 0.018 mmol, 0.5 M solution in CH_2Cl_2) at -20 °C. After stirring for 5 min, the pale yellow reaction mixture was quenched with satd sodium hydrogen carbonate. The aq solution was extracted with CHCl $_3$ and the organic phase was washed with water and brine, and dried ($Na₂SO₄$). Filtration, concentration, and preparative silica gel column chromatography (hexane/AcOEt/ CHCl3, 9:1 to 2:1) afforded a 57 mg of 18 (0.018 mmol, 57%) as a colorless amorphous: $[\alpha]_D^{24}$ –5.0 (c 0.34, CHCl₃), $[\alpha]_D^{24}$ 0.0 (c 0.32, EtOAc); $1H NMR$ (400 MHz, CDCl₃, 0.80:0.20 mixture of rotational isomers) major isomer: 7.48–6.09 (96H, m), 6.32 (0.8H, s), 6.04 (0.8H, d, J=2.2 Hz), 6.01 (0.8H, br s), 5.94 (0.8H, s), 5.84 (0.8H, br s), 5.66 (0.8H, d, $I=2.2$ Hz), 5.59 (0.8H, br s), 5.58–5.53 (0.8H, m), 5.38–3.96 (34.4H, m), 5.16 (0.8H, br s), 3.92 (0.8H, d, $J=11.7$ Hz), 3.81 (0.8H, d, J=11.7 Hz), 3.27-3.14 (1.6H, m); minor isomer: 7.48– 6.09 (24.6H, m), 6.25 (0.2H, s), 6.04–3.83 (10H, m), 3.27–3.14 (0.4H, m); 13 C NMR (100 MHz, CDCl₃, 0.80:0.20 mixture of rotational isomers) major isomer: 165.3, 164.2, 163.8, 158.1, 156.6, 156.5, 156.2, 156.1, 155.9, 155.5, 154.7, 153.5, 152.6, 152.1, 152.0, 149.3, 149.1, 148.9, 148.7, 148.1, 148.0, 143.2, 142.23, 142.17, 138.0– 136.4 (C \times 24), 131.6, 131.2, 130.6, 128.6–125.9 (C \times 50), 125.3, 124.9, 120.2 (2), 119.3, 113.9, 113.8, 113.5, 113.4, 113.0, 111.4, 109.84, 109.76, 109.0, 108.8, 106.2, 105.0, 101.7, 93.8, 93.4, 91.9, 90.6, 77.6, 77.5, 76.1, 75.6, 75.0, 74.9, 74.8, 74.7, 72.9, 71.8-68.9 (C×15), 35.1, 33.8, 26.5; minor isomer was not identified. IR (neat, cm^{-1}) 3090 (w), 3065 (m), 3031 (m), 2936 (m), 2870 (m), 1952 (w), 1860 (w), 1811 (w), 1721 (s), 1593 (s), 1514 (s), 1429 (s), 1373 (s), 1329 (s), 1267 (s), 1217 (s), 1169 (s), 1028 (s), 911 (w), 856 (w); FABMS (m/z) 3236 ($[M+Na]$ ⁺, 23), 3229 (46), 3218 (56), 3214 ($[M+H]$ ⁺, 38), 3196 (39), 3042 (51), 3008 (71), 2958 (77), 2900 (100), 2684 (80), 2598 (64), 2586 (62), 2585 (87), 2425 (77), 2411 (76), 2330 (95), 2293 (83), 2244 (88), 2148 (75), 2108 (73), 2061 (94), 1749 (95), 1611 (100).

4.1.9. [4.8:4".8"]-2.3-cis-3.4-trans:2".3"-cis-3".4"-trans:2"".3"" trans-Dodeca-O-benzyl-(-)-epicatechin-(-)-epicatechin- $(+)$ -catechin-3,3",3""-O-(tri-O-benzyl)gallate (20)

To a solution of 17 (155 mg, 0.072 mmol) and 15 (20.9 mg, 0.018 mmol) in CH_2Cl_2 (40 mL) was added dropwise TMSOTf (0.036 mL, 0.018 mmol, 0.5 M solution in CH_2Cl_2) at -10 °C. After stirring for 5 min, the pale yellow reaction mixture was quenched with satd sodium hydrogen carbonate. The aq solution was extracted with $CHCl₃$ and the organic phase was washed with water and brine, and dried ($Na₂SO₄$). Filtration, concentration, and preparative silica gel column chromatography (hex-AcOEt-CHCl₃, 9:1–2:1) afforded a 44 mg of 19 (0.014 mmol, 76%) as a colorless amorphous: $[\alpha]_D^{26}$ +50.9 (c 0.60, CHCl₃), $[\alpha]_D^{25}$ +46.5 (c 0.62, EtOAc); 1 H NMR (400 MHz, CDCl₃, 0.70:0.30 mixture of rotational isomers) major isomer: 7.48–6.25 (84H, m), 6.41 (0.7H, s), 6.01 (0.7H, d, J=2.2 Hz), 5.95 (0.7H, s), 5.92 (0.7H, br s), 5.75 (0.7H, br s), 5.66 (0.7H, d, J=2.2 Hz), 5.51 (0.7H, br s), 5.26-5.20 (0.7H, m), 5.20 (0.7H, br s), 5.14–5.13 (0.7H, m), 5.13 (0.7H, br s), 5.10–4.46 (25.2H, m), 4.44 (0.7H, d, J=13.9 Hz), 4.36 (0.7H, d, J=12.2 Hz), 4.29 (0.7H, d, J=13.9 Hz), 4.01 (0.7H, d, J=11.5 Hz), 3.94 (0.7H, d, $J=11.5$ Hz), 3.81 (0.7H, d, $J=11.5$ Hz), 3.11 (0.7H, dd, $J=5.6$, 16.9 Hz), 2.87 (0.7H, dd, J=7.5, 17.3 Hz); minor isomer: 7.48-5.58 (36.6H, m), 6.23 (0.3H, s), 6.07 (0.3H, s), 5.26–3.80 (15H, m), 3.14–3.09 $(0.3H, m)$, 2.97–2.90 $(0.3H, m)$; ¹³C NMR $(100 MHz, CDCl₃)$ 0.70:0.30 mixture of rotational isomers) major isomer: 165.3, 164.7, 163.8, 158.1, 156.6, 156.2, 156.09, 156.07, 155.7, 155.5, 154.9, 152.6, 152.5, 152.1, 149.3, 149.1, 148.8 (2), 148.4, 148.2, 143.0, 142.4, 142.1, 138.2–136.4 (C24), 131.7, 131.4, 131.0, 128.7–126.8 $(C \times 54)$, 120.4, 120.1, 119.6, 113.9, 113.7, 113.6, 113.4, 113.0, 111.0, 109.7, 108.9, 108.8, 106.4, 105.1, 101.7, 93.8, 93.6, 92.0, 90.6, 78.3 (2) , 76.0 (2) , 75.5 (2) , 75.1, 74.9 $(\times 2)$, 74.0, 723.9, 71.7–69.1 $(C \times 16)$, 34.4 (4), 33.8 (4), 24.9 (4); minor isomer was not identified. IR (neat, cm⁻¹) 3065 (m), 3032 (m), 2934 (m), 2870 (m), 1952 (w), 1871 (w), 1811 (w), 1720 (s), 1598 (s), 1498 (s), 1429 (s), 1329 (s), 1215 (s), 1119 (s), 1028 (s), 910 (w), 858 (w), 810 (w), 754 (s); FABMS (m/z) 3236 ([M+Na]⁺, 48), 3214 ([M+H]⁺, 19), 3149 (37), 3117 (46), 3073 (55), 3019 (44), 2788 (77), 2713 (79), 2666 (87), 2340 (100), 2244 (75), 1987 (85), 1763 (69), 1725 (78).

4.1.10. [4,8:4",8"]-2,3-cis-3,4-trans:2",3"-cis-3",4"-trans:2"",3""cis-Tri-(-)-epicatechin-3,3",3"" -O-gallate (6)

A solution of 18 (30 mg, 0.0093 mmol) in 22 mL of THF/MeOH/ H₂O, 20:1:1 was hydrogenated over 20% Pd(OH)₂/C (5 mg) for 4 h at rt. Filtration and concentration afforded a pale brown solid, which was purified by Sephadex® LH-20 short column chromatography (MeOH) and HPLC purification to give 10.2 mg of pure 6 (0.0077 mmol, 83%) as an amorphous solid. [α] $_{{\rm D}}^{{\rm 25}}$ +11.3 (c 0.28, acetone) {lit.²⁹ [α] $_{{\rm D}}^{28}$ +13.4 (c 0.93, acetone)}; ¹H NMR (400 MHz, CD₃OD, 0.6:0.4 mixture of rotational isomers, -40 °C) 7.05–5.03 $(17.8H, m)$, 6.18 $(0.6H, s)$, 6.59 $(0.4H, dd, J=3.2, 8.3 Hz)$, 6.50 $(0.4H, d,$ $J=8.3$ Hz), 6.18 (0.4H, s), 6.01 (0.6H, s), 6.00 (0.6H, d, $J=2.2$ Hz), 5.95 $(0.6H, d, J=2.2 Hz), 5.92 (0.6H, s), 5.87 (0.4H, s), 5.75 (0.4H, d,$ $J=2.2$ Hz), 5.72 (0.4H, br s), 5.68 (0.6H, br s), 5.57 (0.4H, br s), 5.27 $(0.4H, m)$, 5.05–5.01 (1H, m), 4.96 (0.6H, d, J=1.7 Hz), 4.81 (0.4H, br s), 4.65 (0.4H, br s), 3.10–2.95 (1H, m), 2.94–2.88 (1H, m); FABMS (m/z) 1346 (21), 1345 ([M+Na]⁺, 93), 1344 (41), 1310 (43), 1265 (51), 1229 (60), 1227 (46), 1170 (53), 1067 (50), 1022 (69), 975 (65), 894 (53), 874 (48), 863 (51), 823 (54), 787 (63), 749 (72), 736 (73); FAB-HRMS calcd for $C_{66}H_{50}O_{30}Na[M+H]^{+}$, 1345.2285; found: 1345.2264.

4.1.11. [4,8:4",8"]-2,3-cis-3,4-trans:2",3"-cis-3",4"-trans:2"",3""trans-(-)-Epicatechin-(-)-epicatechin-(+)-catechin-3,3",3""-Ogallate (7)

A solution of 19 (29 mg, 0.0090 mmol) in 22 mL of THF/MeOH/ H₂O, 20:1:1 was hydrogenated over 20% Pd(OH)₂/C (5 mg) for 7 h at rt. Filtration and concentration afforded a pale brown solid, which was purified by Sephadex[®] LH-20 short column chromatography (MeOH) and HPLC purification to give 6.4 mg of pure 7 (0.0048 mmol, 54%) as an amorphous solid: [α] $_{{\rm D}}^{\rm 25}$ +34.0 (c 0.05, MeOH); 1 H NMR (400 MHz, CD $_3$ OD, 0.55:0.45 mixture of rotational isomers, -40 °C) 7.05-5.25 (19.8H, m), 6.20 (0.55H, s), 5.87 (0.55H, s), 5.73 (0.55H, J=2.2 Hz), 5.70 (0.55H, br s), 5.68 (0.45H, br s), 5.62 (0.55H, br s), 5.56 (0.55H, br s), 5.53 (0.45H, br s), 5.48 (0.55H, br s), 5.50 (0.55H, br s), 4.97 (0.45H, br s), 4.86 (0.55H, br s), 4.74 (0.45H, br s), 4.61 (0.45H, br s), 2.90–2.78 (2H, m); FABMS (m/z) 1346 (20), 1345 ([M+Na]⁺, 32), 1323 ([M+H]⁺, 4.5), 1203 (70), 1202 (40), 1166 (55), 1128 (66), 1064 (29), 985 (55), 970 (50), 955 (49), 899 (44), 845 (43), 840 (53), 757 (100); FAB-HRMS calcd for $C_{66}H_{50}O_{30}Na$ $[M+H]$ ⁺, 1345.2285; found: 1345.2253.

Acknowledgements

Partial financial support of this research under the NOVARTIS Foundation (Japan) for the Promotion of Science is gratefully acknowledged. We also thank the Japan Society for the Promotion of Science (JSPS) for the Young Science Research Fellowship (to A.S.).

References and notes

- 1. Harborne, J. B. The Flavonoids: Advances in Research from 1986; Chapman and Hall: London, 1993.
- 2. Harborne FRS, J. B.; Baxter, H. The Handbook of Natural Flavonoids; John Wiley & Sons: New York, NY, 1999.
- 3. (a) Tückmantel, W.; Kozikowski, A. P.; Romanczyk, L. J. J. Am. Chem. Soc. 1999, 121, 12073–12081; (b) Kozikowski, A. P.; Tückmantel, W.; George, C. J. Org. Chem. 2000, 65, 5371–5381; Kozikowski, A. P.; Tückmantel, W.; Boettcher, G.; Romanczyk, L. J., Jr. J. Org. Chem. 2003, 68, 1641–1658.
- 4. (a) Ohmori, K.; Ushimaru, N.; Suzuki, K. Proc. Natl. Acad. Sci. U.S.A. 2004, 101, 12002–12007; (b) Tarascou, I.; Barathieu, K.; Andre, Y.; Pianet, I.; Dufourc, E.; Fouquet, E. Eur. J. Org. Chem. 2006, 5367–5377; (c) Sharma, P. K.; Kolchinski, A.; Shea, H. A.; Nair, J. J.; Gou, Y.; Romanczyk, L. J., Jr.; Schmitz, H. H. Org. Process Res. Dev. 2007, 11, 422–430; (d) Mohri, Y.; Sagehashi, M.; Yamada, T.; Hattori, Y.; Morimura, K.; Kamo, T.; Hirota, M.; Makabe, H. Tetrahedron Lett. 2007, 48, 5891– 5894; (e) Oyama, K.; Kuwano, M,; Ito, M.; Yoshida, K.; Kondo, T. *Tetrahedron*
Lett. **2008**, 49, 3176–3180; (f) Matthew, A. C.; Bonnet, S. L.; van der Westhuizen, J. H. Org. Lett. 2008, 10, 3865–3868; (g) Mohri, Y.; Sagehashi, M.; Yamada, T.; Hattori, Y.; Morimura, Y.; Hamauzu, K.; Kamo, T.; Hirota, M.; Makabe, H. Heterocycles 2009, 79, 549–563.
- 5. Saito, A.; Nakajima, N.; Tanaka, A.; Ubukata, M. Biosci. Biotechnol. Biochem. 2002, 66, 1764–1767.
- 6. Saito, A.; Nakajima, N.; Tanaka, A.; Ubukata, M. Tetrahedron 2002, 58, 7829–7837.
- 7. Saito, A.; Nakajima, N.; Tanaka, A.; Ubukata, M. Tetrahedron Lett. 2003, 44, 5449–5452.
- 8. Saito, A.; Nakajima, N.; Tanaka, A.; Ubukata, M. Heterocycles 2003, 61, 287–298.
- 9. Saito, A.; Nakajima, N.; Matsuura, M.; Tanaka, A.; Ubukata, M. Heterocycles 2004, 62, 479–489.
- 10. Saito, A.; Tanaka, A.; Ubukata, M.; Nakajima, N. Synlett 2004, 1069–1073.
- 11. Saito, A.; Tanaka, A.; Ubukata, M.; Nakajima, N. Synlett 2004, 2040–2042.
- 12. Saito, A.; Doi, Y.; Matsuura, N.; Tanaka, A.; Ubukata, M.; Nakajima, N. Bioorg. Med. Chem. 2004, 12, 4783-4790.
- 13. Saito, A.; Emoto, M.; Tanaka, A.; Doi, Y.; Shoji, K.; Mizushina, Y.; Ikawa, H.; Yoshida, H.; Matsuura, N.; Nakajima, N. Tetrahedron 2004, 60, 12043–12049.
- 14. Saito, A.; Mizushina, Y.; Ikawa, H.; Yoshida, H.; Doi, Y.; Tanaka, A.; Nakajima, N. Bioorg. Med. Chem. 2005, 13, 2759–2771.
- 15. Sakuda, H.; Saito, A.; Mizushina, Y.; Ikawa, H.; Yoshida, H.; Tanaka, A.; Nakajima, N. Heterocycles 2006, 67, 175–188.
- 16. Yang, C. S.; Maliakai, P.; Meng, X. Annu. Rev. Pharmacol. Toxicol. 2002, 42, 25–54.
- 17. Tachibana, H.; Koga, K.; Fujimura, Y.; Yamada, K. Nat. Struct. Mol. Biol. 2004, 11, 380–381.
- 18. Es-Safi, N. E.; Ghidouche, S.; Ducrot, P. H. Molecules 2007, 12, 2228–2258.
- 19. (a) Kawamoto, H.; Nakatsubo, F.; Murakami, K. J. Wood Chem. Technol. 1990, 10, 59-74; (b) Kawamoto, H.; Nakatsubo, F.; Murakami, K. Mokuzai Gakkaishi 1991, 37, 488–493; (c) Yoneda, S.; Kawamoto, H.; Nakatsubo, F. J. Chem. Soc., Perkin Trans. 1 1997, 1025–1030.
- 20. Obtained NMR signals of **6** and **7** did the broadening, the peak assignment of the ¹³C NMR signal were not possible.
- 21. (a) Blois, M. S. Nature 1958, 181, 1199–1200; (b) Nanjo, F.; Goto, K.; Seto, R.; Suzuki, M.; Sakai, M.; Hara, Y. Free Radical Biol. Med. 1996, 21, 895–902.
- 22. Hatano, T.; Miyatake, H.; Natsume, M.; Osakabe, N.; Takizawa, T.; Ito, H.; Yoshida, T. Phytochemistry 2002, 59, 749–758.
- 23. Kornberg, A.; Baker, T. DNA Replication; Freeman, W.H.: New York, NY, 1992; Chapter 6, pp 197–225.
- 24. Mizushina, Y.; Saito, A.; Tanaka, A.; Nakajima, N.; Kuriyama, I.; Takemura, M.; Takeuchi, T.; Sugawara, F.; Yoshida, H. Biochem. Biophys. Res. Commun. 2005, 333, 101–109.
- 25. These compounds were incubated with each DNA polymerase (0.05 units). One unit of DNA polymerase activity was defined as the amount of enzyme that catalyzed the incorporation of 1 nmol of dNTP (dTTP) into the synthetic DNA template-primers (i.e., $poly(dA)/oligo(dT)₁₂₋₁₈$, $A/T=2/1$) in 60 min at 37 °C under normal reaction conditions for each enzyme.^{26,27} Enzyme activity in the absence of the compound was taken as 100%. Data are expressed as the mean \pm SD; $n=3$
- 26. Mizushina, Y.; Tanaka, N.; Yagi, H.; Kurosawa, T.; Onoue, M.; Seto, H.; Horie, T.; Aoyagi, N.; Yamaoka, M.; Matsukage, A.; Yoshida, S.; Sakaguchi, K. Biochim. Biophys. Acta 1996, 1308, 256–262.
- 27. Mizushina, Y.; Yoshida, S.; Matsukage, A.; Sakaguchi, K. Biochim. Biophys. Acta 1997, 1336, 509–521.
- 28. Foo, J. Y.; Karchesy, J. J. Phytochemistry 1991, 30, 667–670.
- 29. Kashiwada, Y.; Nonaka, G.; Nishioka, I. Chem. Pharm. Bull. 1986, 34, 4083–4091.